

Q- The measurements of a photoelectric-effect experiment are graphed in in Figure P38.44, in which the intervals along the horizontal and vertical axes are respectively given by  $5.9 \times 10^{14}$  Hz and 2.50 V, respectively.

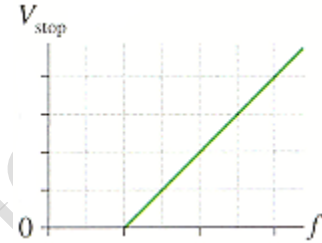
(a) What is the work function of the cathode?

If the stopping potential is zero means the electrons are having no kinetic energy and need not to be stopped. As the stopping potential graph starts from  $f = 2 \times 5.9 \times 10^{14}$  Hz before that the electron is not emitted or  $2 \times 5.9 \times 10^{14}$  Hz is the threshold frequency  $f_0$  and hence the work function is given by

$$\phi = hf_0 = 6.63 \times 10^{-34} \times 2 \times 5.9 \times 10^{14} = 7.82 \times 10^{-19} \text{ J}$$

Or  $\phi = 4.89 \text{ eV}$

Hence the work function of the cathode is **4.89 eV**



(b) What experimental value of Planck's constant is obtained from these data?

The Einstein's equation can be written in terms of frequency  $f$  and stopping potential as

$$hf - \phi = eV_{stop}$$

Or  $V_{stop} = \frac{hf}{e} - \frac{\phi}{e}$

The equation is of the form  $y = mx + c$  and  $h/e$  can be given as the slope of  $V_{stop}, f$  curve hence

$$h = e \times \text{slope of } V_{stop}, f \text{ curve}$$

or  $h = 1.6 \times 10^{-19} \times (4 \times 2.5) / (4 \times 5.9 \times 10^{14}) = 6.78 \times 10^{-34} \text{ J.s}$

Thus, the experimental value of planks constant is  **$6.78 \times 10^{-34} \text{ J s}$**