

Q- Kr-85 is a radioactive isotope that is used in measuring paper thickness. It decays by emission with a half-life of 10.8 years

- Write down the decay equation showing the daughter isotope and decay products of this process.
- A Kr-85 source had activity 30 mCi when new. Estimate its activity after 5 years' use.
- What is the activity of a 30 mCi source in becquerels?
- What is the mass of Kr-85 in the source when new?

(a) We can write the formula for the beta decay of Krypton-85 as:



The Krypton-85 nucleus turns into a nucleus of the element Rubidium which still has a sum of 85 protons and neutrons, and a beta particle (electron) flies out with (687 keV) energy. In this case, a *gamma ray* photon has energy 513,990 electron volts (514 keV)

(b) Activity $dN/dt = 30 \text{ m Ci} = 30 \times (3.7 \times 10^7 \text{ disintegrations/sec}) = 1.11 \times 10^9 \text{ dis/s}$

Decay constant $\lambda = 0.693/T = 0.693/10.8 = 0.0642 \text{ year}^{-1}$

The activity varies with time as

$$A = A_0 e^{-\lambda t}$$

gives activity after five years

$$\begin{aligned} A &= (1.11 \times 10^9 \text{ dis/s}) e^{-(0.0642 \text{ per year}) \times (5 \text{ year})} = 1.11 \times 10^9 \times 0.7254 \\ &= 8.0522 \times 10^8 \text{ dis/s} = 21.76 \text{ mCi.} \end{aligned}$$

(c) 1 becquerel = 1 Bq = 1 decay per second

$$30\text{mCi} = 1.11 \times 10^9 \text{ dis/s} = 1.11 \times 10^9 \text{ Bq.}$$

(d) $DN/dt = -N_0 \lambda$ gives numerically (-ve because N is decreasing)

$$\begin{aligned} N_0 &= (dN/dt)/\lambda = (1.11 \times 10^9 \text{ dis/s})/0.0642 \text{ year}^{-1} \\ &= 1.11 \times 10^9 \times 365 \times 24 \times 3600 = 3.50 \times 10^{16} \\ &= 3.50 \times 10^{16}/6.023 \times 10^{23} \text{.mol} = 5.81 \times 10^{-8} \text{ mol.} \end{aligned}$$

Therefore, mass of sample = $5.81 \times 10^{-8} \times 85 = 4.94 \times 10^{-6} \text{ gm.}$